

## Tocopherols and flavonoids of SOS-7 halophyte

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### RESUMEN

#### Tocopheroles y flavonoides de halofito SOS-7

Halofito es una semilla oleaginosa codificada como SOS-7 (semilla oleaginosa *Salicornia*, séptimo año de selección). Los tocoferoles del aceite de halofito SOS-7 fueron determinados directamente en el aceite usando cromatografía líquida de alta presión acoplada a detector fluorescente. Se encontró que el aceite contenía 710 ppm de tocoferoles totales. Los tocoferoles alfa, beta, gamma y delta, se encontraron a niveles de 38.2, 1.0, 58.7 y 2.1%, respectivamente.

Nueve glicósidos flavonoides fueron aislados e identificados de las semillas y se encontró que pertenecen a la clase flavonol dentro de los flavonoides. Estos flavonoles fueron identificados como: quercetina-3,7-diglucosido, quercetina-3-glucosido-7-galactosido, quercetina-3-sophorosido, quercetina-3-glucosido, quercetina-3-galactosido, isorannetina-3, 7-di-glucosido, isorannetina-3-glucosido, kampferol-3, 7-diglucosido y kampferol-3-glucosido.

**PALABRAS-CLAVE:** Aceite de halofito SOS-7 - Flavonoide - Tocopherol.

### SUMMARY

#### Tocopherols and flavonoids of SOS-7 halophyte.

Halophyte is an oil seed coded as SOS-7 (*Salicornia* Oil Seed, 7<sup>th</sup> year of selection). Tocopherol constituents of SOS-7 halophyte oil were determined directly in the oil by using high pressure liquid chromatography coupled to fluorescence detector. It was found that the oil contains 710 ppm total tocopherols. The tocopherol constituents, alpha, beta, gamma and delta, were found at the level of 38.2, 1.0, 58.7 and 2.1% respectively.

Nine flavonoid glycosides were isolated and identified from the seeds and it was found that they belong to the flavonol class of flavonoids. These flavonol compounds were identified as: quercetin-3, 7-diglucoside, quercetin-3-glucoside-7-galactoside, quercetin-3-sophoroside, quercetin-3-glucoside, quercetin-3-galactoside, isorhamnetin-3, 7-di-glucoside, isorhamnetin-3-glucoside, kaempferol-3, 7-diglucoside and kaempferol-3-glucoside.

**KEY-WORDS:** Flavonoid - SOS-7 halophyte oil - Tocopherol.

## 1. INTRODUCTION

The world-wide demand for vegetable oils has greatly increased, this led to the search for new seed oils. SOS-7 Halophyte, is a new oil seed crop, as climatic tolerant, has been successfully tried in desert areas and irrigated with sea water (1). In continuation to the previous studies in this

laboratory (2) (3) (4) on characterizing halophyte, SOS-7 oil produced from locally cultivated halophyte plant, it was advisable to carry out further studies dealing with tocopherols and flavonoids to complete the picture of the seed constituents.

Tocopherol content of an oil is of equal interest to fatty acid composition in evaluating the quality of the oil. Tocopherols are the best-known and most universally distributed antioxidants, constitute fat-soluble vitamin E. Four forms, alpha, beta, gamma and delta tocopherols (alpha-T, beta-T, gamma-T and delta-T) (Fig. 1) are found in vegetable oils.

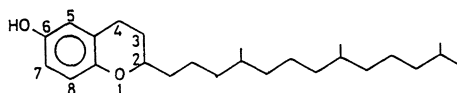


Fig.1

Structure of tocopherols: 5,7,8 trimethyl (alpha-T); 5,8 dimethyl (beta-T); 7,8 dimethyl (gamma-T) and 8 methyl (delta-T).

Gamma-T has been found to be a more effective antioxidant than beta-T which is in turn effective than alpha-T (5).

The term flavonoid is applied to compounds with a structure based on flavone (2-phenyl chromone). Flavone consists of two benzene rings (A and B) joined together by a gamma pyrone ring (Fig. 2). The structures, commonly found in nature, are flavan-3 ols (catechins), flavonols, flavanones, flavanonols and flavan-3,4-diols.

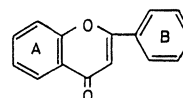


Fig.2

Flavonoid compounds are widely distributed in higher plants. They were isolated from different parts of plants with variations in the types of compounds found according to various anatomical tissues of any plant (6). It has been reported that flavonoids have antioxidant effect on unsaturated fatty acids (7). As far as one knows, nothing has been

reported about the flavonoids of SOS-7 halophyte in the current literature.

## 2. MATERIALS AND METHODS

### 2.1. Tocopherols (T)

The oil was extracted from the ground seeds using  $\text{CHCl}_3/\text{MeOH}$  (2:1, v/v) and the solvent was evaporated under vacuum. Standard tocopherols (St.T), alpha-T, beta-T, gamma-T and delta-T were purchased from E. Merk, Germany.

### High pressure liquid Chromatography (HPLC)

Tocopherol constituents were determined directly in the oil using Toyo Soda-CCPM, high pressure liquid chromatograph, coupled to Hitachi-650-10-S fluorescence detector with an excitation and emission wave lengths of 290 and 325 nm respectively. The analysis was carried out under the following operating conditions: column, 150 x 4.6 mm packed with silica gel YMC-A-012 SIL; solvent, hexane: isopropyl alcohol (100: 0.5 v/v) at flow rate of 1-2 ml/min; sample size, 20 microlitre (1-1.5% oil in hexane) and chart speed, 5mm/min (8). The amounts of the resolved tocopherol components were estimated from the measured areas under the appropriate peaks (PA) in comparison with those of the standard tocopherol solutions (St.T) according to the following equation (8) (9) (10).

$$T.\text{ppm/Kg oil} = \frac{\text{P.A. of resolved T} \times \text{g oil/ml} \times 1000}{\text{P.A. of St.T.} \times \text{mg/ml (St.T)}}$$

### 2.2. Flavonoids

The solvents, n-hexane, chloroform, ethanol, methanol and water were successively used in extracting the different constituents from the ground seeds. The extracts were concentrated under reduced pressure and applied on Whatman paper No. 1 to determine their flavonoid constituents. A two dimensional chromatographic technique was used with solvent systems n-butanol/acetic acid/water (BAW 4:1:5, v/v/v) and 15% acetic acid in the first and second directions respectively. Paper chromatography showed the absence of flavonoids in n-hexane and chloroform extracts (negative results with  $\text{AlCl}_3$  spray reagent). The other extracts gave positive results with  $\text{AlCl}_3$  spray reagent and they all had the same flavonoid compounds. The three extracts (ethanol, methanol and water) were combined and concentrated under reduced pressure. The concentrated extract was applied on a cellulose column, and eluted with ethanol containing increasing amounts of water. The collected fractions were further separated into nine single components by applying elution technique on Whatman paper No.3. The isolated compounds were inves-

tigated according to the standard methods of identification (11) (12) (13) (14).

All the authentic samples (markers) of the flavonol compounds used in this paper were separated from the Egyptian flora and identified by the researchers in the laboratory of Plant Taxonomy and Egyptian Flora, National Research Centre, Cairo. Standard sugars were purchased from BDH Chemicals Ltd., England.

## 3. RESULTS AND DISCUSSION

### 3.1. Tocopherols

Tocopherols are important antioxidants which also function as vitamin E. It is known that tocopherols are superior radical chain-breaking antioxidants (15) compared to synthetic ones. Several methods have been developed for the analysis of tocopherols in vegetable oils. It was found that HPLC analysis of tocopherols is a sensitive and selective method. In the present work, HPLC analysis of crude SOS-7 halophyte oil, using a fluorescence detector indicates that this oil is rich in tocopherols and contains as much as 710 ppm total tocopherols. Four tocopherol isomers, alpha-T (38.2%), beta-T (1.0%), gamma-T (58.7%) and delta-T (2.1%) were detected. Their relative retention times were 3.35, 5.65, 6.0 and 10.4 min respectively. From these results it can be seen that SOS-7 halophyte oil is rich in gamma-T which is superior as antioxidant (5).

### 3.2. Flavonoids

All the nine isolated compounds appeared as brown spots under UV light which changed to yellow on exposure to ammonia vapour. Complete acid hydrolysis of the nine compounds (1-9) gave quercetin, kaempferol and isorhamnetin as aglycones together with glucose and galactose as sugar moieties which means that these compounds belong to the O-flavonol glycosides. Acid hydrolysis also showed that compounds 1, 2, 3, 6 and 8 are disaccharide glycosides while compounds 4, 5, 7 and 9 are monosaccharide glycosides. The  $R_f$  values, chemical and UV data of the isolated compounds are recorded in Tables I, II and III respectively.

#### 3.2.1. Disaccharide Glycosides (Compounds 1, 2, 3, 6 and 8)

On complete acid hydrolysis compound 1 and 3 gave quercetin and glucose while compound 2 gave quercetin, glucose and galactose. Compound 6 gave isorhamnetin and glucose while compound 8 gave kaempferol and glucose (Table II).

UV data (Table III) show that compounds 1, 2, 6 and 8 were occupied in positions 3 & 7. Occupation of position 3 was confirmed by the stability of the UV spectrum after addition of NaOMe. However, position 7 was confirmed to

Table I  
R<sub>f</sub>-values of the isolated flavonol glycosides

Compounds (*)	R <sub>f</sub> -values X 100 (**)			
	BAW	H <sub>2</sub> O	15%	PhOH
1. Q37G	17	41	45	43
2. Q3G7Gal	13	51	51	35
3. Q3Soph	33	30	43	45
4. Q3G	47	13	31	53
5. Q3Gal	44	13	29	55
6. I37G	23	41	51	76
7. I3G	54	28	35	70
8. K37G	25	54	51	67
9. K3G	61	14	35	60

(\*) Q37G = Quercetin-3,7-diglucoside.  
Q3G7Gal = Quercetin-3-glucoside-7-galactoside.  
Q3Soph = Quercetin-3-sophoroside.  
Q3G = Quercetin-3-glucoside.  
Q3Gal = Quercetin-3-galactoside.  
I37G = Isorhamnetin-3,7-di-glucoside.  
I3G = Isorhamnetin-3-glucoside.  
K37G = Kaempferol-3,7-diglucoside.  
K3G = Kaempferol-3-glucoside.

(\*\*) BAW = n-Butanol:Acetic acid:Water (4:1:5).  
15% = 15% Acetic acid.  
PhOH = Phenol:Water (80:20).

Table II  
Chemical data of the isolated compounds

Compounds	Complete acid hydrolysis (*)		Mild acid hydrolysis
	Aglycone	Sugar	
1. Q37G	Q	G	Q7G
2. Q3G7Gal	Q	G, Gal	Q7Gal
3. Q3 Soph	Q	G	Q3G
4. Q3G	Q	G	—
5. Q3Gal	Q	Gal	—
6. I37G	I	G	I7G
7. I3G	I	G	—
8. K37G	K	G	K7G
9. K3G	K	G	—

(\*) Q = Quercetin. I = Isorhamnetin. K = Kaempferol.  
G = Glucose. Gal = Galactose.

be occupied by the absence of the shift in band II with sodium acetate compared with that of methanol and also with the absence of a small peak at 320 nm with NaOMe.

The structures of these compounds were further confirmed by carrying out mild acid hydrolysis which gave intermediates (Table II), namely, quercetin-7-galactoside, quercetin-3-glucoside, isorhamnetin-7-glucoside and kaempferol-7-glucoside. These intermediates were co-chromatographed with authentic samples using different solvent systems namely, n-butanol: acetic acid: water (BAW) (4:1:5); 15% acetic acid and phenol: water (PhOH) (80:20). It was found that the R<sub>f</sub> values of the isolated compounds (Table I) were identical with those reported in the literature for the disaccharide flavonoids.

Table III  
U.V. data of the isolated compounds

Compounds	MeOH	NaOMe	AlCl <sub>3</sub>	AlCl <sub>3</sub> -HCl	NaOAc	NaOAc-H <sub>3</sub> BO <sub>3</sub>
1. and 2	257	268sh	267	267	265	262
Q37G	265sh	275	297sh	297sh	295sh	380
and	357	417	362	357	360	
Q3G7Gal			415	400	420sh	
3.	255sh	280	255sh	255sh	270	265
Q3Soph	267	325sh	270	270	278sh	278sh
	278sh	420	360	360	295sh	295sh
	357		405sh	405sh	365	375
4. and 5.	257	268sh	262	262	270	262
Q3G	265	282	300sh	300sh	370	380
and	357	335sh	357sh	357sh		
Q3Gal		425	420	400		
6.	255	250sh	270	269	257	257
I37G	270sh	280	305sh	305sh	270sh	270sh
	350	430	364	364	368	360
			405	405	420	
7.	255	272	270	268	275	258
I3G	267sh	327	300sh	300sh	315	366sh
	295sh	415	356	358	388	305sh
	355		405	400		360
8.	265	245	274	274	265	265
K37G	318sh	270	300	290sh	318sh	320sh
	350	295sh	355	350	360	350
		390	400	395	400sh	
9.	266	275	276	276	276	256
K3G	300sh	325	305	305	310	285sh
	350	400	355	355	390	345
			400	400		

Further investigation was carried out on compound 3 by H<sub>2</sub>O<sub>2</sub> oxidation which gave rise to a disaccharide. This disaccharide was confirmed to be sophorose after co-chromatographed with authentic sugar samples.

From the above data it can be concluded that compounds 1, 2, 3, 6 and 8 could be identified as quercetin-3,7-diglucoside, quercetin-3-glucoside-7-galactoside, quercetin-3-sophoroside, isorhamnetin, 3-7-diglucoside and kaempferol-3-7-diglucoside.

### 3.2.2. Monosaccharide Glycosides (Compounds 4, 5, 7 and 9)

The results of the chemical investigation, complete and mild acid hydrolysis, of compounds 4, 5, 7 and 9 are recorded in Table II. It can be noticed that glycosides 4 and 5 gave quercetin as an aglycone and glucose and galactose, respectively, as sugar moieties on complete acid hydrolysis. However, compounds 7 and 9 gave isorhamnetin and kaempferol as aglycone moieties respectively and glucose as sugar moiety. The types of the aglycones and sugars were confirmed by co-chromatographing the products of hydrolysis with authentic samples using different solvent systems (Table II).

Table III shows the UV data of compounds 4, 5, 7 and 9 and the data are in agreement with those reported for the flavonol monosaccharide glycosides.

It can be concluded that monosaccharide glycosides 4, 5, 7 and 9 can be identified as quercetin-3-glucoside, quercetin-3-galactoside, isorhamnetin-3-glucoside and kaempferol-3-glucoside.

From all the results discussed above, it can be generally concluded that SOS-7 halophyte is rich in natural antioxidants namely, tocopherols and flavonoids.

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