

## Fatty acid composition of the pollen lipids of *Cycas revoluta* Thunb

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**SUMMARY:** The fatty acid (FA) composition of total extractable and non extractable with chloroform lipids of *C. revoluta* pollen was determined. Among other minor FAs, unusual  $\Delta 5$  polymethylene-interrupted FA,  $\Delta 5, 11$ -octadecadienoic acid was found. This FA was found in the seed lipids of *C. revoluta* earlier, but it was discovered for the first time in pollen lipids.

**KEYWORDS:** *Cycas revoluta*; Fatty acids; Gymnosperm species; Pollen; Sago palm;  $\Delta 5$ -fatty acids

**RESUMEN:** *Composición en ácidos grasos de los lípidos del polen de palmeras Cycas revoluta.* Se determinó la composición en ácidos grasos (AG) de los lípidos totales extraíbles y no extraíbles con cloroformo del polen de la palmera *C. revoluta*. Entre otros ácidos grasos menores se encontró un AG  $\Delta 5$  inusual, el ácido octadecadienoico,  $\Delta 5, 11$ -polimetilen-interrumpido. Este AG ya fue descrito en los lípidos de semillas de *C. revoluta*, pero en los lípidos del polen es la primera vez que se describen.

**PALABRAS CLAVE:** Ácidos grasos; Ácidos grasos  $\Delta 5$ -; *Cycas revoluta*; Especies de gimnospermas; Palma de sagú; Polen

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### 1. INTRODUCTION

*Cycas* L. (false sago palm) is an ancient genus of Gymnosperms, a group of more than 90 species, the only genus of the family Cycadaceae. It is expected that Cycadaceae are the earliest seed plants, like the ginkgo, descending from long extinct seed ferns (Laubenfels and Adema, 1998). Cycadaceae are pollinated by the wind similar to other dioecious plants. For decorative purposes only a few *Cycas* species are used, among which the most popular is *Cycas revoluta*, originally from Southeast Asia. The lifetime of *C. revoluta* reaches more than 100 years, greenhouse specimens can

bloom only a few times during growth (Hill *et al.*, 2004; Kramer and Green, 1990). However this phenomenon is hardly to be called flowering as such, because *Cycas* plants belong to the varieties of Gymnosperms, which have no fruits and no true flowers. The specimen of *C. revoluta*, which has grown in greenhouse of the K.A. Timiryazev Institute of Plant Physiology for several decades, has bloomed this year for the first time. Since the fatty acid (FA) composition of *Cycas* lipids has been investigated only in leaves (Mongrand *et al.*, 2001) and seeds (Takagi and Itabashi, 1982), and taking into account the extreme rarity of the flowering of this plant, we decided to study the

FA composition of the total lipids of its pollen (microspores).

## 2. MATERIALS AND METHODS

### 2.1. Plant material and extraction of lipids

Pollen grains were collected by means of glass rods from the microstrobile of a blooming male plant of the *Cycas revoluta* Thunb. growing in the greenhouse of the K.A. Timiryazev Institute of Plant Physiology of RAS. Lipids were extracted from the plant material (500 mg) with 100 mL of purified chloroform for 30 min, the extract was filtered out and the solvent was removed with a vacuum evaporator. The FAs of the lipids of the dry residue were converted into methyl esters by direct transesterification with 5 mL of a 10% methanolic solution of acetyl chloride (Sidorov *et al.*, 2014). The FAs of non extractable residue were obtained by direct saponification with 10 mL of a 6% solution of potassium hydroxide in 80% aqueous methanol followed by extraction of free FAs with hexane and their subsequent conversion into methyl esters by a common procedure (Sidorov *et al.*, 2014). All the solvents contained 0.001% of butylated hydroxytoluene as antioxidant.

### 2.2. Synthesis of 4'4'-dimethyloxazoline derivatives of fatty acids

In order to determine the double bond position of unidentified FAs, we used mass-spectrometry of their 4'4'-dimethyloxazoline derivatives also known as DMOXes, which we synthesized following a common procedure with slight modifications. We added 200  $\mu$ L of oxalyl chloride to 5 mg of free fatty acids obtained by saponification of plant material (see section 2.1), then placed a screw cap vial with this mixture into the 50 ml centrifuge tube filled with anhydrous sodium sulfate. The tube was left for one hour in the water bath heated to 45 °C with the subsequent evaporation of oxalyl chloride under a stream of argon. Then we added 300  $\mu$ L of a 20% solution of amino-2-propanol in dichloromethane to the chloroanhydrides of the FFAs and left the vial for one hour at the room temperature. The solvent was evaporated under a stream of argon. Afterwards we added 300  $\mu$ L of trifluoroacetic anhydride to the dry residue and kept the vial at 45 °C for 1 hour. Finally we removed the excess solvent in a stream of argon, added 100  $\mu$ L of distilled water and 200  $\mu$ L of hexane, shook the vial vigorously, collected the hexane layer into a new vial and dried it over anhydrous sodium sulfate for 20 min (Christie, 2012b). The DMOXes thus obtained were immediately analyzed by GC–MS (see section 2.3).

### 2.3. Analysis of fatty acids

The qualitative and quantitative FA compositions in the lipid preparations were determined by GC–MS using an internal standard technique; heptadecanoic acid methyl ester was an internal standard (Sidorov *et al.*, 2014). Fatty acid methyl esters were analyzed by GC–MS using an Agilent 7890A GC device fitted with a capillary column (DB-23, 60 m  $\times$  0.25 mm) containing a grafted (50% cyanopropyl)-methylpolysiloxane polar liquid phase as a 0.25  $\mu$ m-thick film. The FAMES were separated under the following conditions: operational gas (helium) flow in the column at 1 mL/min, sample volume, 1  $\mu$ L; flow split ratio, 1:10; evaporator temperature, 260 °C. The oven temperature program was as follows: from 130 to 170 °C at 6.5 °C/min, to 215 °C at 2.75 °C/min (25 min at this temperature), to 240 °C at 40 °C/min, and 50 min at 240 °C, operational temperature of the mass selective detector (Agilent 5975C MSD), 240 °C; ionization energy, 70 eV. For identifying individual FAME species and calculating their concentrations in the mixture, a NIST research library v. 2.0 and MSD Chem Station E.02.00.493 software were used (Sidorov *et al.*, 2014). All experiments were performed with three replicates.

## 3. RESULTS AND DISCUSSION

The FA compositions of total extractable and non extractable with  $\text{CHCl}_3$  (mainly neutral and polar) lipids of *C. revoluta* pollen are presented in Table 1. One can see that the diversity of the FA composition was higher in the extractable with  $\text{CHCl}_3$  pollen lipids (28 individual FAs species) than in the non extractable ones (21 FAs species). Major FAs in both pollen lipid fractions were palmitic (16:0), oleic ( $\Delta$ 9-18:1) and linoleic ( $\Delta$ 9,12-18:2) acids; stearic (18:0) and  $\alpha$ -linolenic ( $\Delta$ 9,12,15-18:3) acids were also present in appreciable quantities. In both the extractable and non extractable with  $\text{CHCl}_3$  lipids of *C. revoluta* pollen in minor quantities several unusual FAs namely  $\Delta$ 7-18:1,  $\Delta$ 9,11-18:2,  $\Delta$ 5,9-18:2 (taxoleic),  $\Delta$ 5,9,12-18:3 (pinolenic), and  $\Delta$ 5,11,14-20:3 (sciadonic) were found as well as several FAs with very long chains ( $\text{C}_{\geq 20}$ ).

During the analysis of the  $\text{CHCl}_3$ -extractable lipids of *C. revoluta* pollen, we drew the attention to a minor FA (RT of its methyl ester was equal to 21.86 min, RRT relative to  $\text{C}_{18:0}$ =1.063 and calculated ECL equal to 18.47). An automated search of mass spectra libraries NIST08 and Wiley did not result in the identification of the monitoring component because of more than 90% overlap with library mass spectrum of the methyl ester of the taxoleic acid ( $\Delta$ 5,9-18:2). However, since the chromatographic parameters calculated for this peak differed from the peak of the latter, we decided to identify unknown

TABLE 1. Fatty acids composition of extractable and non extractable with  $\text{CHCl}_3$  lipids of *C. revoluta* pollen, *mas.*-% of total FAs<sup>a</sup>

Fatty acid	ECL	Extractable lipids	Non extractable lipids
14:0	14.00	0.4±0.0	0.6±0.2
15:0	15.00	0.2±0.0	0.4±0.0
16:0	16.00	25.2±0.4	30.9±1.0
Δ7-16:1	16.20	0.3±0.0	0.4±0.0
Δ9-16:1	16.29	0.3±0.0	0.2±0.0
Δ7,10-16:2	16.68	0.1±0.0	– <sup>b</sup>
Δ7,10,13-16:3	17.27	0.1±0.0	–
17:0	17.00	0.3±0.0	1.0±0.0
18:0	18.00	2.4±0.1	6.3±0.1
Δ7-18:1	18.15	0.1±0.0	0.6±0.0
Δ9-18:1	18.29	27.7±0.1	18.6±0.3
Δ11-18:1	18.33	1.0±0.0	1.2±0.0
Δ5,9-18:2	18.41	0.8±0.0	0.4±0.0
X	18.47	0.8±0.0	–
Δ9,11-18:2	18.65	0.1±0.0	–
Δ9,12-18:2	18.79	30.7± 0.5	28.4±0.1
Δ5,9,12-18:3	18.90	0.7±0.0	–
Δ9,12,15-18:3	19.32	3.7±0.2	3.1±0.0
19:0	19.00	0.1±0.0	0.1±0.1
20:0	20.00	0.1±0.1	1.3±0.2
Δ11-20:1	20.25	0.4±0.0	–
Δ8,11-20:2	20.40	0.4±0.0	–
Δ11,14-20:2	20.75	0.1±0.0	0.1±0.1
Δ5,11,14-20:3	20.91	2.6±0.1	1.9±0.1
21:0	21.00	0.1±0.0	0.1±0.1
22:0	22.00	0.7±0.0	2.9±0.4
23:0	23.00	0.1±0.0	0.1±0.1
24:0	24.00	0.4±0.0	1.3±0.3

<sup>a</sup>means ± SD.<sup>b</sup>“–” – not detected.

FA (X, presumably the  $x,y$ -18:2) using the mass spectrometry of its 4,4-alkenyl-dimethyloxazoline (DMOX) derivative.

The mass spectrum of the DMOX derivative of unknown FA from pollen lipids of *C. revoluta* is presented in Figure 1. Molecular ion  $M^+$  with  $m/z = 333$  and its fragmentation profile indicates the location of an octadecadienoic acid with ethylene bonds in an unusual place. The existence of a fragmentary ion with an odd value  $m/z=153$  (marked with asterisk) is an important diagnostic sign indicating the position of the double bond at the 5th carbon atom of the FA residue. A characteristic couple of fragmentary ions with a difference of 26 *a.m.u.* for the Δ5 double bond location with  $m/z$  140 and 166 (Christie, 2012a) is also present in the mass spectrum. From Figure 1 one can see that the relative intensity of ions with greater masses is significantly

lower than the  $m/z=153$  ion intensity, except for fragmentary ion with  $m/z=180$ , which usually indicates that the double bonds in the FA acyl are separated by more than one  $-\text{CH}_2-$  group. The presence of this ion is characteristic for the FA with so-called polymethylene-interrupted *bis*-oriented double bonds (Christie, 2012a; Wolff and Christie, 2002). The character of fragmentation, leading a pair of fragmentary ions with  $m/z=153$  and 180 to be generated, is shown in the Figure 1.

This assumption is confirmed by the fact that the mass-spectra of DMOX derivatives of other FAs with the same value of  $M^+$ , for example, Δ6- or Δ9-octadecenoic, as well as the Δ8,9-methylene-9-heptadecenoic or Δ13-cyclopentyl-2-enyl-tridecanoic acids, are substantially different in their pattern of fragmentation (Christie, 2012b). Starting the ion with  $m/z=166$ ,

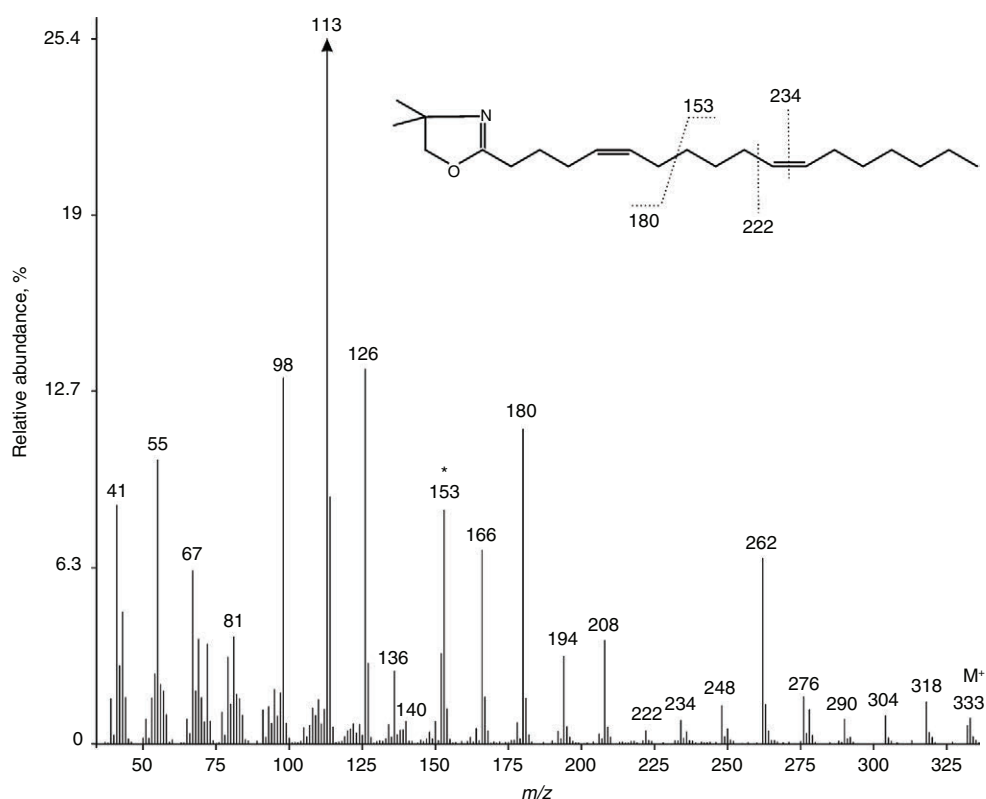


FIGURE 1. Mass spectrum of the DMOX derivative of unknown FA (X, see Table) from pollen lipids of *C. revoluta*. Intensities of the remaining fragmentary ions are brought relatively to ion with  $m/z=113$ , one of the characteristic ions of FA DMOX derivatives, taken as 100%.

a series of fragmentary ions, differing in 14 *a.m.u.* follows. This is typical for fragmentation under the sequential detachment of  $-\text{CH}_2-$  groups from FA acyl. This order is broken by ions with  $m/z$  222 and 234, where the difference in 12 *a.m.u.* indicates the presence of an ethylene bond at the 11<sup>th</sup> carbon atom of FA acyl (a second pair of ions, important to determine the exact location of the double bond must differ by 26 *a.m.u.*, in this case, it is ions with  $m/z$  222 and 248). Thus, it can be concluded that the unusual fatty acid found in the total lipids of *C. revoluta* pollen is nothing other than  $\Delta 5,11$ -octadecadienoic acid, belonging to a group of bis-polymethylene-interrupted FAs (UPIFAs).

Earlier, the FA composition of *C. revoluta* lipids was studied in the leaves (Mongrand *et al.*, 2001) and seeds (Takagi, Itabashi, 1982) of this plant. However, to the best of our knowledge, the FA composition of *C. revoluta* pollen lipids has never been investigated. Leaf lipids were found to contain 21  $\text{C}_{14}$ - $\text{C}_{22}$  FAs (Mongrand *et al.*, 2001), and the seed ones more than 23  $\text{C}_{13}$ - $\text{C}_{22}$  FAs (Takagi and Itabashi, 1982). The major FAs of leaf lipids were 16:0,  $\Delta 9,12$ -18:2 and  $\Delta 9,12,15$ -18:3 acids and 16:0,  $\Delta 9$ -18:1 and  $\Delta 9,12$ -18:2 acids predominated in the seed ones. Both leaf and seed lipids of *C. revoluta* contained small quantities of several  $\text{C}_{18}$  and  $\text{C}_{20}$   $\Delta 5$ -UPIFA,

however, besides pollen,  $\Delta 5,11$ -18:2 FA was found only in the latter (Takagi and Itabashi, 1982). The composition of  $\Delta 5$ -UPIFA in the leaves and seeds was more diverse than in pollen lipids (6, 6, and 4 individual FA species, respectively). In particular, the leaves and seeds contained  $\Delta 5,9,12,15$ -18:4 and  $\Delta 5,11,14,17$ -20:4 FAs, which were absent in the pollen.

Takagi and Itabashi have studied the FA composition of the seed lipids of 21 species of Gymnosperms, but besides *C. revoluta*  $\Delta 5,11$ -18:2 FA was identified only in the seed lipids of *G. biloba*, *Ephedra sinica* and in *Podocarpus macrophylla* (Takagi and Itabashi, 1982). According to data of Mongrand *et al.*, who studied the FA composition of the leaf lipids of 137 species of Gymnosperms belonging to 14 families, including the leaves of *C. revoluta*, lipids of the latter, as the overwhelming majority of other Gymnosperms studied, along with conventional  $\text{C}_{14}$ - $\text{C}_{22}$  FAs in small quantities contained *anteiso*-17:0 FA, taxoleic, coniferonic ( $\Delta 5,9,12,15$ -18:4), pinolenic,  $\Delta 5,11$ -20:2, sciadonic and uniperonic ( $\Delta 5,11,14,17$ -20:4) acids (Mongrand *et al.*, 2001). However, in the leaves of all the Gymnosperm species studied by Mongrand *et al.*  $\Delta 5,11$ -18:2 FA was not detected.

#### 4. CONCLUSIONS

To the best of our knowledge,  $\Delta 5,11$ -octadecadienoic acid was found for the first time in *Dictyostelium discoideum* lipids by Davidoff and Korn (Davidoff and Korn, 1962). Soon afterwards, Gellerman and Schlenk discovered this FA in the seed and leaf lipids of *Ginkgo biloba* (Gellerman and Schlenk, 1963) but unambiguously the structure of this FA has been established by Wolff and coworkers, who found it in *Ephedra* as well as in *G. biloba* seed oils and identified by it GC-MS of its nicotinyl and DMOX derivatives (Wolff *et al.*, 1999). Wolff and coworkers proposed a trivial name – ephedrinic acid for  $\Delta 5,11$ -18:2 FA, and suggested several possible pathways for its biosynthesis (Wolff and Christie, 2002; Wolff *et al.*, 1999). The first of the proposed pathways was connected with the  $C_2$ -elongation of palmitoleic acid with the formation of *cis*-vacenic acid and its subsequent  $\Delta 5$ -desaturation. The authors also suggested that there are two  $\Delta 5$ -desaturases, strictly specific to oleat- and *cis*-vacenate (Wolff *et al.*, 1999). Under the second probable pathway of the  $\Delta 5$ -UPIFA biosynthesis the crucial role is played not by the positions of the double bonds in the FA acyl, but by the carbon chain length. This hypothesis was based on the results of the investigation of the FA composition of numerous species of Gymnosperms (Wolff and Christie, 2002; Wolff *et al.*, 2000).

Wolff and coworkers also supposed the simultaneous existence of two  $\Delta 5$ -desaturases specific for  $C_{18}$  and  $C_{20}$  unsaturated FAs or for the  $\Delta 9$  and  $\Delta 11$  positions of the first double bond, respectively, because along with ephedrinic acid, the  $\Delta 5,11$ -eicosadienoic (keteleonic) acid often appeared (Wolff and Christie, 2002; Wolff *et al.*, 1999; Wolff *et al.*, 2000). Nevertheless, the question of the biosynthesis of unusual  $\Delta 5$ -acids in Gymnosperms remains unanswered.

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